Arctic methane sources: Isotopic evidence for atmospheric inputs

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[1] By comparison of the methane mixing ratio and the carbon isotope ratio ($\delta^{13}C_{CH4}$) in Arctic air with regional background, the incremental input of CH₄ in an air parcel and the source $\delta^{13}C_{CH4}$ signature can be determined. Using this technique the bulk Arctic CH₄ source signature of air arriving at Spitsbergen in late summer 2008 and 2009 was found to be -68%, indicative of the dominance of a biogenic CH₄ source. This is close to the source signature of CH₄ emissions from boreal wetlands. In spring, when wetland was frozen, the CH₄ source signature was more enriched in 13 C at $-53 \pm 6\%$ with air mass back trajectories indicating a large influence from gas field emissions in the Ob River region. Emissions of CH₄ to the water column from the seabed on the Spitsbergen continental slope are occurring but none has yet been detected reaching the atmosphere. The measurements illustrate the significance of wetland emissions. Potentially, these may respond quickly and powerfully to meteorological variations and to sustained climate warming. Citation: Fisher, R. E., et al. (2011), Arctic methane sources: Isotopic evidence for atmospheric inputs, Geophys. Res. Lett., 38, L21803, doi:10.1029/2011GL049319.

1. Introduction

[2] Arctic CH₄ emissions may have played a major role both in modern CH₄ excursions [Dlugokencky et al., 2011] and in past global climatic change [Nisbet and Chappellaz, 2009]. Arctic CH₄ comes from varied sources, most of which respond quickly to temperature change, with strong positive feedbacks such that warming feeds warming. Four of the five warmest decades of a 2000-year-long reconstruction of Arctic temperatures occurred between 1950 and 2000 [Kaufman et al., 2009]. An increase in global CH₄ in 2007 following years of near stability [Dlugokencky et al., 2009, 2011; Rigby et al., 2008] may have been in part a response to the Siberian heatwave that summer. Northern wetland emissions are likely to increase sharply with temperature [Bohn et al., 2007] and may have helped drive past

global glacial/interglacial changes [Nisbet and Chappellaz, 2009; Petrenko et al., 2009].

2. Isotopic Signature of Sources of Methane to the Arctic

[3] Arctic CH₄ sources can be identified by isotopic signature (Table 1). In the Eurasian Arctic, emissions from wetlands sampled in ambient air give a characteristic and consistent isotopic signature with $\delta^{13}C_{CH4}$ in the range –69 to -65%. Emissions occur from May melt to October freezeup, and may show an exponential (Arrhenius) increase with temperature [Nisbet, 1989; Westermann and Ahring, 1987]. Keeling plot experiments by our group reported elsewhere (S. Sriskantharajah, manuscript in preparation, 2011) find a relatively constant source $\delta^{13}C_{CH4} - 68.5 \pm 0.7\%$ for wetland emissions from Northern Finland during the summer. Warming permafrost and thermokarst can emit CH₄, made by methanogens from organic matter. The isotopic composition of CH₄ from thermokarst lakes in E. Siberia ranges widely, from -83% to -58% [Walter et al., 2006]. Fire is another CH₄ source likely to increase with warming and drought. With warm dry summers, such as in Russia in 2010, tundra fires may become more frequent [*Qiu*, 2009]. We report here new results of source studies that have been carried out to identify the isotopic source signature of CH₄ from forest fires in Canada. Source signatures were identified using Keeling plots of CH₄ measured in ambient air collected close to prescribed burns. The mean isotopic composition of CH₄ emitted by a boreal jack pine forest fire in Ontario, Canada (46°47'N, 83°20'W) in May 2007 was $-27.8 \pm 0.3\%$. CH₄ from a mature pine fire in Northwest Territories, Canada (61°21'N, 117°40'W) in June 2010 had a mean isotopic composition of $-28.7 \pm 0.7\%$. These results fall within the range measured for pine forest fire emissions in the United States (-30 to -21%) [Chanton et al., 2000]. Fire-sourced CH₄, $\delta^{13}C_{CH4}$ around -28% for northern pine forests, is isotopically very distinct from wetland CH₄. Increased burning, also fingerprinted by elevated CO, should enrich CH₄ in ¹³C.

[4] Industrial gas leaks are a major component of global anthropogenic CH₄ emissions, with many of the world's largest gas fields being north of the Arctic circle. The giant gas fields around the Ob estuary supply much of Europe's heat and electric power via long pipelines, and Arctic gas leaks may have contributed in part to the rapid rise in the global CH₄ burden in the 1980s [*Dlugokencky et al.*, 1994]. Gas can escape to the atmosphere both from leaks in wells and pipes, and from normal technical operations [*Reshetnikov et al.*, 2000]. CH₄ from the W. Siberian fields varies isotopically. W Siberian gas measured by the Uni-

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Table 1. δ^{13} C in Northern Methane Sources

Source	Reference	δ ¹³ C (‰)
Wetland, N. Finland:	S. Sriskantharajah	-68.5 ± 0.7
summer spring thaw autumn freeze-up	(unpublished data)	-66.3 ± 0.6 -64.9 ± 4.0
Wetland, Hudson Bay Lowlands, Canada	Kuhlmann et al. [1998]	-60 ± 3
Tundra, Alaska	Quay et al. [1988]	-73 to -55
Wetland, Siberia	Nisbet [2005]	-67 ± 2
Ebullition from thermokarst lakes, N. Siberia	Walter et al. [2006]	−83 to −58
W Siberian natural gas	Cramer et al. [1999]	-51 ± 3
Marine clathrate, W. Spitsbergen	This work	-50 ± 5
Onshore hydrate, Mackenzie delta, Canada	Lorenson et al. [1999]	−48.7 to −39.6
Pine forest fires, Canada	This work	-28 ± 1

versity of Heidelberg [Nisbet, 2001] and leaked gas in average ambient air in the production region around Korotchaevo tower in the Ob River region measured by our group [Nisbet, 2005] has an isotopic signature of $-51 \pm 2\%$, which is comparable to other measurements of W Siberian production gas [Cramer et al., 1999].

[5] CH₄ emitted by decomposing Arctic gas hydrates is also isotopically variable, depending on the original source of the CH₄ trapped in hydrate, or free gas pooled below it, and also on fractionation that may occur on release to the atmosphere. Hydrate is a capacitor, trapping CH₄, not a source per se. Thermogenic gas from deep geological carbon stores is relatively rich in ¹³C. In contrast, gas sourced from biological methanogenesis at shallow levels is richer in ¹²C. Measurements of CH₄ from decomposing CH₄ hydrate show a wide isotopic range: hydrate gas with $\delta^{13}C_{CH4}$ –62 to –56‰ has previously been measured in the Norwegian Arctic [Milkov, 2005]; and \sim -72 to -66% in the gas in the Nyegga pockmarks [Vaular et al., 2010]. Extensive Arctic CH₄ hydrates are vulnerable to decomposition with warming [Nisbet, 1989], if shallow submarine and terrestrial clathrates destabilise. There is evidence for strong CH₄ ebullition to the atmosphere in response to the warming of shallow offshore sediments of the East Siberian Arctic Shelf in the far eastern Arctic [Shakhova et al., 2010a, 2010b]. In deeper water, CH₄ bubble plumes have recently been identified from the shelf edge west of Spitsbergen [Westbrook et al., 2009].

[6] We report here (Table 2) $\delta^{13}C_{CH4}$ source signatures of $-55 \pm 2\%$ and $-46 \pm 3\%$ in samples of CH₄ hydrate in two sediment cores collected from within the gas hydrate stability zone, during a cruise along the continental slope NW of Spitsbergen in 2008 [Westbrook et al., 2009]. CH₄ in any bubbles that escape from the sea-surface to the atmosphere may be further enriched in ¹³C following partial oxidation of gas in the water column or sediments, because methano-

trophs would preferentially consume the lighter carbon isotope. Isotopic fractionation of CH₄ in the water column has been observed in the nearby Spitsbergen continental shelf [Damm et al., 2005].

3. Experimental Methods

[7] To identify major emissions of Arctic CH₄ and assess their relative inputs to the Arctic CH₄ budget, ambient air samples were collected daily at the Zeppelin station, Spitsbergen, during late summer/autumn 2008, spring 2009 and late summer/autumn 2009. Zeppelin (78°54′N, 11°53′E), at 475 m above sea level is a representative background site normally above the planetary boundary layer with limited influence from local CH₄ sources [*Pedersen et al.*, 2005]. Simultaneously with the summer 2008 campaign, in the nearby Fram Strait, off W. Spitsbergen, intensive air sampling was carried out on the bridge of RRS *James Clark Ross*. Shipboard sampling was repeated during two cruises with RV *Jan Mayen* over the gas plume field in the Fram Strait in July 2009 and October 2010.

[8] Air was collected by pumping into 5 litre tedlar bags. Mixing ratio and δ^{13} C of atmospheric CH₄ was measured in all air samples in the Dept. of Earth Sciences at Royal Holloway University of London. For air samples collected at Zeppelin in Autumn 2009 and shipboard samples from 2010 CH₄ mixing ratios were measured using a Picarro Cavity Ringdown spectrometer (CRDS) with a repeatability of ± 0.3 ppb. For all other air samples CH₄ mixing ratios were determined using an HP 5890 gas chromatograph (GC) with a flame ionization detector (FID) with a repeatability of ±5 ppb. The GC and CRDS instruments are calibrated for CH₄ in the range 1831 to 1965 ppb using NOAA air standards and all CH₄ mixing ratio data are given on the NOAA04 scale [Dlugokencky et al., 2005]. CH₄ δ^{13} C was analysed using a modified gas chromatography isotope ratio mass spectrometry (GC-IRMS) system (Trace Gas and Isoprime mass spectrometer, Isoprime Ltd.) with 0.05% repeatability [Fisher et al., 2006]. All measurements were made in triplicate. Isotope ratios are given in δ -notation on the VPDB (Vienna Pee Dee Belemnite) scale.

4. Results and Discussion

[9] During the summer 2008 cruise, more than 250 plumes of gas bubbles were identified [Westbrook et al., 2009] using a 38 kHz sonar. Plumes were identified in the same area in 2009 (Figure 1). In seawater samples collected in 2008 dissolved CH₄ concentrations were up to 20 times greater at the sea bottom than in the surface waters above the plumes. Some bubble plumes rose close to the sea surface. Despite the proximity of the plumes beneath the ship's track, air collected on the ship did not contain elevated CH₄ compared with contemporaneous samples from Zeppelin (Figure 2), which is located 70 km NW of the observed

Table 2. Hydrate Samples Collected in the Fram Strait in September 2008^a

Core ID	Location	Seawater Depth (m)	Depth of Hydrate Below Sediment Surface (cm)	δ ¹³ C _{CH4} (‰)
JR211-33-GC	Plume field 78°41.07'N, 08°16.36'E	890 m	>126 cm	-54.6 ± 1.7
JR211-26-GC	Vestnesa ridge 79°00.39'N, 06°54.26'E	1210 m	>193 cm	-45.7 ± 2.7

^aMethane δ^{13} C was measured in 3 chips from each sample of hydrate.

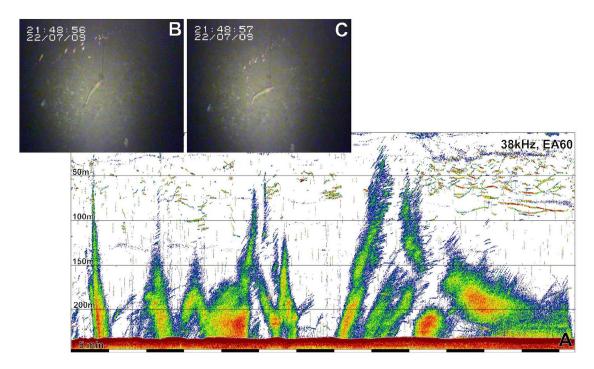


Figure 1. (a) Simrad ER 60 echosounder (38 kHz) showing several flares at an active seep site on the west Spitsbergen slope in 240 m water depth. Horizontal axis shows bars for 5 minute intervals during positioning of the ship for CTD sampling. Distance along the x axis is non-linear as the ship was drifting: for a transect through the plume field on the W. Spitsbergen continental margin see *Westbrook et al.* [2009]. (b and c) Frame grabs of the video survey at the sea floor around this seep and the larger bubbles that contribute to the flares in the echogram. Bubbles visible above the fish in Figures 1b and 1c are about 8 to 10 mm diameter.

plume field. Nor were the shipboard results significantly different isotopically from Zeppelin samples. Probably, complete CH₄ dissolution and gas stripping [McGinnis et al., 2006; Rehder et al., 2009] or methanotrophic oxidation occur as the bubbles rise. Oceanic CH₄ and oxygen measurements following the 2010 Deepwater Horizon oil spill in the Gulf of Mexico [Kessler et al., 2011] showed that rapid methanotrophic oxidation of released CH₄ occurred in the water column. Our results show that seabed CH₄ emissions from much shallower waters, though demonstrably present in the sonar records and water column measurements, did not reach the atmosphere.

[10] Bulk CH₄ inputs to Arctic air may be assessed by sampling air masses from a range of directions, using high precision $\delta^{13}C_{CH4}$ measurement, if isotopic signatures of regional sources are known. The y-intercept of a Keeling plot of $\delta^{13}C_{CH4}$ against 1/CH₄ mixing ratio in the shipboard and Zeppelin air samples during the period 23 August to 20 September 2008 gives the bulk source input of $\delta^{13}C_{CH4}$ $-67.5 \pm 1.1\%$. This bulk result is strikingly similar to typical wetland emissions, though with the caution that it may record a mix of disparate sources. Additional shipboard campaigns in summer and autumn 2009 and autumn 2010 produced similar results, with CH₄ mixing ratios measured in the gas plume area no higher than those measured at the Zeppelin station. The source signature of CH₄ in these samples was $\delta^{13}C_{CH4}$ -68.6 ± 4.5% in July 2009 and -68.7 ± 4.4‰ in October 2010. CH₄ in air sampled daily at the Zeppelin station in September to October 2009 had a $\delta^{13}C_{CH4}$ source signature of -67.4 ± 3.1% (Table 3).

Destruction of CH_4 by OH is small in the Arctic where OH concentrations are low so there is little isotopic enrichment by reaction with OH over the periods considered. The large observed variations in CH_4 concentration are related to meteorological dynamics in the surface boundary layer with a timescale much shorter than the lifetime of tropospheric CH_4 and the bulk isotopic signature reflects the CH_4 source mix.

[11] The measured bulk input signature at Spitsbergen of \sim -68‰ for 2008–10 shows the late summer CH₄ input is dominantly from δ^{13} C depleted sources such as wetland (δ^{13} C_{CH4} typically -69‰), or thermokarst (-83 to -58‰) (Table 1). If the mean biogenic source signature is -69 or -70‰, then a simple mass balance equation implies no more than 10% of the -68‰ Arctic summer input to Spitsbergen air can be from isotopically heavier inputs such as gas leaks (circa -51‰), fires, or local clathrate identified on the W Spitsbergen slope (-55 to -46‰).

[12] Any isotopically heavy CH_4 input from boreal fires was small as no large CO excursions were observed during the period. CO mixing ratio measured semi-continuously at the Zeppelin station by the Norwegian Institute for Air Research (NILU) using an in situ RGA3 (Reduction Gas Analyser, Trace Analytical) was compared with the CH_4 mixing ratio measured in the bags collected at the Zeppelin station at the same time. There is a weak correlation ($R^2 = 0.49$) between excess CO and CH_4 at Zeppelin during the period 09 to 20 September 2008, with a $CO:CH_4$ ratio of 0.3 ppb(CO)/ppb(CH_4). This is much smaller than the ratio expected from boreal forest fires which would produce

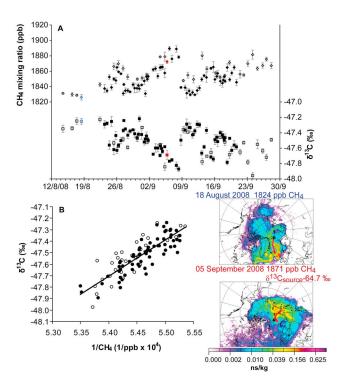


Figure 2. (a) Summer 2008 time series of CH_4 mixing ratio (diamonds) and $\delta^{13}C_{CH4}$ (squares) in ambient air at the Zeppelin station (hollow) and on the RRS James Clark Ross in the Fram Strait (filled). Error bars denote one standard deviation in triplicate measurements of each air sample. (b) Corresponding Keeling plot. The y axis intercept of the Keeling plot is $-67.5 \pm 1.1\%$. FLEXPART footprint emission sensitivity plots using ECMWF wind fields are shown for air arriving at Zeppelin from the Atlantic on 18 August (blue points on Figure 2a) and from Northern Siberia on 05 September (red points on Figure 2a).

CO and CH₄ with a CO:CH₄ ratio of between 12 and 38 ppb/ppb [Cofer et al., 1998]. Given the proximity of giant Russian Ob River gasfields, which are known to emit at least some leakage in summer [Reshetnikov et al., 2000], some part of the isotopically heavier contribution to the bulk mix must come from gas fields.

[13] In contributing to the $\delta^{13}C_{CH4}$ –68‰ signature of the bulk CH₄ increment measured at Zeppelin, the total local clathrate input during the sampling period is likely to have been very small. Note however this does not exclude inputs from isotopically lighter shallow-sourced CH₄ emitted from clathrates further afield such as in the eastern Arctic. Our results from the high Arctic are consistent with the finding in 1999 in mid-latitude boreal wetland along the Trans-Siberian railroad and Ob river, that $\delta^{13}C_{\text{source}} = -62.9 \pm 0.7\%$, which suggested CH₄ from wetlands dominated the substantial CH₄ excess in western Siberia [*Tarasova et al.*, 2006].

[14] In spring, when Arctic wetland is still frozen, the CH₄ in air samples is significantly more enriched in ^{13}C (Figure 3). In March to May, the bulk Arctic source signature, calculated from measurements in air samples collected daily at the Zeppelin station, gives an Arctic springtime source with $\delta^{13}C_{CH4}$ –52.6 \pm 6.4%. This difference from summer is consistent with observations in the Hudson Bay

Lowlands where boreal wetland has a sharp seasonal onset of emissions in June and a seasonal switch-off in September [Pickett-Heaps et al., 2010]. Most likely the dominant Arctic sources in winter and spring are gas field emissions. Some (aseasonal) clathrate-derived contribution to this relatively heavy springtime signature is also possible, but given the summer result, local clathrate sources are likely to be small. More remote clathrate sources may contribute if flaw polynas (ice free regions) allow release of CH₄ from shallow clathrate to the Arctic atmosphere during the winter. However, in this spring study very few air masses arrived at Zeppelin with trajectories from known emission areas on the East Siberian Arctic Shelf region [Shakhova et al., 2010a, 2010b].

[15] Footprint emission sensitivities for air arriving at Zeppelin and at the ship were calculated using the Lagrangian particle dispersion model, FLEXPART [Stohl et al., 1998]. The highest CH₄ mixing ratios were measured in air masses from Siberia. The atmospheric transport on a twenty-day timescale shows that the Zeppelin observatory is highly sensitive to surface emissions in the Arctic and particularly to emissions in high latitude Eurasia in winter [Hirdman et al., 2010]. Emissions that led to the highest observed CH₄ mixing ratios in the summer 2008 study period (04–08 September) were from northern Siberia, crossing a large area from the Ob River to Eastern Siberia, including the East Siberian Arctic Shelf. The mean source signature, $\delta^{13}C_{CH4}$, for these specific days was $-65 \pm 3\%$. Assuming only wetland (-69%) and gas leaks (-51%) are responsible for this CH₄, then this signature may be apportioned to 78% wetlands and 22% gas leak, but there may also be significant input from East Siberian Arctic Shelf emissions [Shakhova et al., 2010b] with variable δ^{13} C. Air from Canada and Greenland also had high CH₄ content (19th to 20th September) with an isotopic source signature of -69‰, indicating Canadian wetland sources dominated [Kuhlmann et al., 1998].

5. Summary and Conclusions

[16] The results imply the dominant Arctic summer CH₄ source in 2008 and 2009 was biogenic, from wetland. This is consistent with evidence from Siberia of the importance of wetland CH₄ [*Tarasova et al.*, 2009]. In winter, gas emissions dominate the CH₄ input. Submarine emissions along the West Spitsbergen slope currently input negligible CH₄ to the air in summer, despite the clear evidence for gas plumes in the water column. However, this could change rapidly if a warming Atlantic warms the West Spitsbergen

Table 3. Methane Source Signatures Calculated From Daily or Twice Daily Sampling at the Zeppelin Station (Ny-Ålesund, Spitsbergen) and in the Fram Strait^a

Sampling Site	Dates	$\delta^{13}C_{CH4}$ (‰)
Zeppelin Zeppelin Zeppelin Fram Strait Fram Strait Fram Strait	14th August–14th October 2008 6th March–9th May 2009 5th September–4th October 2009 23rd August–20th September 2008 21st July–26th July 2009 9th October–28th October 2010	-68.7 ± 2.4 -52.6 ± 6.4 -67.4 ± 3.1 -66.9 ± 1.3 -68.6 ± 4.5 -68.7 ± 4.4

^aThe source signatures were given by the y-axis intercept of a Keeling plot of the data from each period listed, using a geometric mean regression.

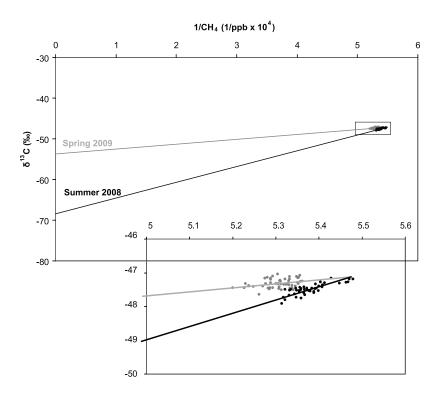


Figure 3. Keeling plot of methane mixing ratio and δ^{13} C in ambient air sampled at the Zeppelin station during August to October 2008 (black) and March to May 2009 (grey). Y axis intercepts are $-68.7 \pm 2.4\%$ ($R^2 = 0.65$) in summer/autumn 2008 and $-52.6 \pm 6.4\%$ (low correlation, $R^2 = 0.14$) in spring 2009.

current [Westbrook et al., 2009]. Gas hydrates are widespread in thick sediments in the Fram Strait between Spitsbergen and Greenland. If the sea bottom warms, the gas hydrate stability zone will move further down the continental slope. Given the steep slopes, earthquakes triggered by ice-melt unloading could produce submarine landslides, triggering further emissions [Berndt et al., 2009].

[17] Wetland CH₄ emissions respond rapidly to warming, such that the warming can feed the warming [Nisbet and *Ingham*, 1995], as evidenced by their importance in glacial terminations [Nisbet and Chappellaz, 2009]. In particular, Arctic and boreal wetlands are likely to respond immediately to sustained heatwaves and increases in precipitation. Fire CH₄ is also more likely with elevated temperatures. There is a strong need for more regular CH₄ isotopic measurements in the high Arctic and intercomparison with the isotopic data from flasks currently collected at Barrow [Miller et al., 2002], Alert [Dlugokencky et al., 2009; Nisbet, 2005] and Zeppelin to measure bulk inputs of CH₄ to the Arctic. Isotopic data can then be used to constrain emissions in both regional and global inversion models [Bousquet et al., 2006]. High frequency, ideally continuous, monitoring of CH₄ δ^{13} C_{CH4} from a number of Arctic sites, onshore and offshore, will be important if future changes in Arctic sources are to be quantified.

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References

Berndt, C., S. Brune, E. Nisbet, J. Zschau, and S. V. Sobolev (2009), Tsunami modeling of a submarine landslide in the Fram Strait, *Geochem. Geophys. Geosyst.*, 10, Q04009, doi:10.1029/2008GC002292.

Bohn, T. J., D. P. Lettenmaier, K. Sathulur, L. C. Bowling, E. Podest, K. C. McDonald, and T. Friborg (2007), Methane emissions from western Siberian wetlands: Heterogeneity and sensitivity to climate change, *Environ. Res. Lett.*, 2(4), 045015, doi:10.1088/1748-9326/2/4/045015.

Bousquet, P., et al. (2006), Contribution of anthropogenic and natural sources to atmospheric methane variability, *Nature*, 443(7110), 439–443, doi:10.1038/nature05132.

Chanton, J. P., C. M. Rutkowski, C. C. Schwartz, D. E. Ward, and L. Boring (2000), Factors influencing the stable carbon isotopic signature of methane from combustion and biomass burning, *J. Geophys. Res.*, 105(D2), 1867–1877, doi:10.1029/1999JD900909.

Cofer, W. R., E. L. Winstead, B. J. Stocks, J. G. Goldammer, and D. R. Cahoon (1998), Crown fire emissions of CO₂, CO, H₂, CH₄, and TNMHC from a dense jack pine boreal forest fire, *Geophys. Res. Lett.*, 25(21), 3919–3922, doi:10.1029/1998GL900042.

Cramer, B., H. S. Poelchau, P. Gerling, N. V. Lopatin, and R. Littke (1999), Methane released from groundwater: The source of natural gas accumulations in northern West Siberia, *Mar. Pet. Geol.*, *16*(3), 225–244, doi:10.1016/S0264-8172(98)00085-3.

Damm, E., A. Mackensen, G. Budeus, E. Faber, and C. Hanfland (2005), Pathways of methane in seawater: Plume spreading in an Arctic shelf environment (SW-Spitsbergen), *Cont. Shelf Res.*, 25(12–13), 1453–1472, doi:10.1016/j.csr.2005.03.003.

Dlugokencky, E. J., K. A. Masarie, P. M. Lang, P. P. Tans, L. P. Steele, and E. G. Nisbet (1994), A dramatic decrease in the growth rate of atmospheric methane in the northern hemisphere during 1992, *Geophys. Res. Lett.*, 21(1), 45–48, doi:10.1029/93GL03070.

Dlugokencky, E. J., R. C. Myers, P. M. Lang, K. A. Masarie, A. M. Crotwell, K. W. Thoning, B. D. Hall, J. W. Elkins, and L. P. Steele (2005), Conversion of NOAA atmospheric dry air CH₄ mole fractions to a gravimet-

- rically prepared standard scale, *J. Geophys. Res.*, 110, D18306, doi:10.1029/2005JD006035.
- Dlugokencky, E. J., et al. (2009), Observational constraints on recent increases in the atmospheric CH₄ burden, *Geophys. Res. Lett.*, 36, L18803, doi:10.1029/2009GL039780.
- Dlugokencky, E. J., E. G. Nisbet, R. Fisher, and D. Lowry (2011), Global atmospheric methane in 2010: Budget, changes and dangers, *Philos. Trans. R. Soc. A*, 369(1943), 2058–2072.
- Fisher, R., D. Lowry, O. Wilkin, S. Sriskantharajah, and E. G. Nisbet (2006), High-precision, automated stable isotope analysis of atmospheric methane and carbon dioxide using continuous-flow isotope-ratio mass spectrometry, *Rapid Commun. Mass Spectrom.*, 20(2), 200–208, doi:10.1002/rcm.2300.
- Hirdman, D., H. Sodemann, S. Eckhardt, J. F. Burkhart, A. Jefferson, T. Mefford, P. K. Quinn, S. Sharma, J. Strom, and A. Stohl (2010), Source identification of short-lived air pollutants in the Arctic using statistical analysis of measurement data and particle dispersion model output, Atmos. Chem. Phys., 10(2), 669–693, doi:10.5194/acp-10-669-2010.
- Kaufman, D. S., et al. (2009), Recent warming reverses long-term Arctic cooling, *Science*, 325(5945), 1236–1239, doi:10.1126/science.1173983.
- Kessler, J. D., et al. (2011), A persistent oxygen anomaly reveals the fate of spilled methane in the deep *Gulf of Mexico*, *Science*, 331(6015), 312–315, doi:10.1126/science.1199697.
- Kuhlmann, A. J., D. E. J. Worthy, N. B. A. Trivett, and I. Levin (1998), Methane emissions from a wetland region within the Hudson Bay Lowland: An atmospheric approach, *J. Geophys. Res.*, 103(D13), 16,009–16,016.
- Lorenson, T. D., M. J. Whiticar, A. Waseda, S. R. Dallimore, and T. S. Collett (1999), Gas composition and isotopic geochemistry of cuttings, core and gas hydrate from the JAPEX/JNOC/GSC Mallik 2L-38 gas hydrate research well, Geol. Surv. Can. Bull., 544, 143–164.
- McGinnis, D. F., J. Greinert, Y. Artemov, S. E. Beaubien, and A. Wuest (2006), Fate of rising methane bubbles in stratified waters: How much methane reaches the atmosphere?, J. Geophys. Res., 111, C09007, doi:10.1029/2005JC003183.
- Milkov, A. V. (2005), Molecular and stable isotope compositions of natural gas hydrates: A revised global dataset and basic interpretations in the context of geological settings, *Org. Geochem.*, 36(5), 681–702, doi:10.1016/j.orggeochem.2005.01.010.
- Miller, J. B., K. A. Mack, R. Dissly, J. W. C. White, E. J. Dlugokencky, and P. P. Tans (2002), Development of analytical methods and measurements of C-13/C-12 in atmospheric CH₄ from the NOAA Climate Monitoring and Diagnostics Laboratory global air sampling network, J. Geophys. Res., 107(D13), 4178, doi:10.1029/2001JD000630.
- Nisbet, É. G. (1989), Some northern sources of atmospheric methane: Production, history and future implications, *Can. J. Earth Sci.*, 26(8), 1603–1611, doi:10.1139/e89-136.
- Nisbet, E. G. (Ed.) (2001), Russian emissions of atmospheric methane: Study of sources, Final Rep. INTAS 97-2055, Eur. Commiss., Brussels, Belgium.
- Nisbet, E. G. (Ed.) (2005), Meth-MonitEUr: Methane monitoring in the European Union and Russia, Final Rep., Sect. 6, EC - contract EVK2-CT-2002-00175, Eur. Commiss., Brussels, Belgium, Accessible (quick view) via www.nilu.no/methmoniteur/files/MethMonFinalSectio6.pdf.
- Nisbet, E. G., and J. Chappellaz (2009), Shifting gear, quickly, *Science*, 324(5926), 477–478, doi:10.1126/science.1172001.
- Nisbet, E. G., and B. Ingham (1995), Methane output from natural and quasi-natural sources: A review of the potential for change and for biotic and abiotic feedbacks, in *Biotic Feedbacks in the Global Climate System:* Will the Warming Feed the Warming?, edited by G. M. Woodwell and F. T. Mackenzie, pp. 188–218. Oxford Univ. Press. Oxford.
- F. T. Mackenzie, pp. 188–218, Oxford Univ. Press, Oxford. Pedersen, I. T., K. Holmen, and O. Hermansen (2005), Atmospheric methane at Zeppelin Station in Ny-Alesund: Presentation and analysis of in situ measurements, *J. Environ. Monit.*, 7(5), 488–492, doi:10.1039/bd16034d
- Petrenko, V. V., et al. (2009), ¹⁴CH₄ measurements in Greenland ice: Investigating last glacial termination CH₄ sources, *Science*, *324*(5926), 506–508, doi:10.1126/science.1168909.
- Pickett-Heaps, C. A., D. J. Jacob, K. J. Wecht, E. A. Kort, S. C. Wofsy, G. S. Diskin, D. E. J. Worthy, J. O. Kaplan, I. Bey, and J. Drevet

- (2010), Magnitude and seasonality of wetland methane emissions from the Hudson Bay Lowlands (Canada), *Atmos. Chem. Phys. Discuss.*, 10, 22,415–22,435, doi:10.5194/acpd-10-22415-2010.
- Qiu, J. (2009), Tundra's burning, *Nature*, 461(7260), 34–36, doi:10.1038/461034a.
- Quay, P. D., S. L. King, J. M. Lansdown, and D. O. Wilbur (1988), Isotopic composition of methane released from wetlands: implications for the increase in atmospheric methane, *Global Biogeochem. Cycles*, 2, 385–397, doi:10.1029/GB002i004p00385.
- Rehder, G., I. Leifer, P. G. Brewer, G. Friederich, and E. T. Peltzer (2009), Controls on methane bubble dissolution inside and outside the hydrate stability field from open ocean field experiments and numerical modeling, *Mar. Chem.*, 114(1–2), 19–30, doi:10.1016/j.marchem.2009.03.004.
- Reshetnikov, A. I., N. N. Paramonova, and A. A. Shashkov (2000), An evaluation of historical methane emissions from the Soviet gas industry, *J. Geophys. Res.*, 105(D3), 3517–3529.
- Rigby, M., et al. (2008), Renewed growth of atmospheric methane, *Geophys. Res. Lett.*, 35, L22805, doi:10.1029/2008GL036037.
- Shakhova, N., I. Semiletov, I. Leifer, A. Salyuk, P. Rekant, and D. Kosmach (2010a), Geochemical and geophysical evidence of methane release over the East Siberian Arctic Shelf, *J. Geophys. Res.*, 115, C08007, doi:10.1029/2009JC005602.
- Shakhova, N., I. Semiletov, A. Salyuk, V. Yusupov, D. Kosmach, and O. Gustafsson (2010b), Extensive methane venting to the atmosphere from sediments of the East Siberian Arctic Shelf, *Science*, 327(5970), 1246–1250, doi:10.1126/science.1182221.
- Stohl, A., M. Hittenberger, and G. Wotawa (1998), Validation of the Lagrangian particle dispersion model FLEXPART against large-scale tracer experiment data, *Atmos. Environ.*, 32(24), 4245–4264, doi:10.1016/S1352-2310(98)00184-8.
- Tarasova, O. A., C. A. M. Brenninkmeijer, S. S. Assono, N. F. Elansky, T. Rockmann, and M. Brass (2006), Atmospheric CH₄ along the Trans-Siberian railroad (TROICA) and river Ob: Source identification using stable isotope analysis, *Atmos. Environ.*, 40(29), 5617–5628, doi:10.1016/j.atmosenv.2006.04.065.
- Tarasova, O. A., S. Houweling, N. Elansky, and C. A. M. Brenninkmeijer (2009), Application of stable isotope analysis for improved understanding of the methane budget: Comparison of TROICA measurements with TM3 model simulations, *J. Atmos. Chem.*, 63(1), 49–71, doi:10.1007/s10874-010-9157-v.
- Vaular, E. N., T. Barth, and H. Haflidason (2010), The geochemical characteristics of the hydrate-bound gases from the Nyegga pockmark field, Norwegian Sea, Org. Geochem., 41(5), 437–444, doi:10.1016/j. orggeochem.2010.02.005.
- Walter, K. M., S. A. Zimov, J. P. Chanton, D. Verbyla, and F. S. Chapin (2006), Methane bubbling from Siberian thaw lakes as a positive feedback to climate warming, *Nature*, 443(7107), 71–75, doi:10.1038/nature05040.
- Westbrook, G. K., et al. (2009), Escape of methane gas from the seabed along the West Spitsbergen continental margin, *Geophys. Res. Lett.*, *36*, L15608, doi:10.1029/2009GL039191.
- Westermann, P., and B. K. Ahring (1987), Dynamics of methane production, sulfate reduction, and denitrification in a permanently waterlogged alder swamp, *Appl. Environ. Microbiol.*, 53(10), 2554–2559.
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